

## Class overview today - December 3, 2018

- Part I Basic concepts of thermochronology
  - Basic concepts of thermochronology
  - Estimating closure temperatures
- Part II Low-temperature thermochronology (online only)
  - Definition of low-temperature thermochronology
  - Three common low-temperature thermochronometers
- Part III Quantifying erosion with thermochronology (online only)
  - Basic concepts of heat transfer as a result of erosion
  - Estimation of exhumation rates from thermochronometers



#### Introduction to Quantitative Geology Lesson 6.2

#### Low-temperature thermochronology

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• Define low-temperature thermochronology

- Introduce three common types of low-temperature thermochronometers
  - Helium dating (The (U-Th)/He method)
  - Fission-track dating (The FT method)
  - Argon dating (The <sup>40</sup>Ar/<sup>39</sup>Ar method)



## What is low-temperature thermochronology?

#### Low-T thermochronology uses thermochronometers with effective closure temperatures below ~300°C

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Intro to Quantitative Geology

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# What is low-temperature thermochronology?



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Ehlers and Farley, 2003

- Thermochronometer ages provide a constraint on the time-temperature history of a rock sample
  - In many cases, the age is the time since the sample cooled below the system-specific effective closure temperature



Ehlers and Farley, 2003

 Because the temperatures to which thermochronometers are sensitive generally occur at <u>depths of I to >I5 km</u> and <u>ages are</u> <u>typically I to 100's of Ma</u>, they record long-term cooling through the upper part of the crust and can be used to calculate <u>long-term average rates of tectonics and erosion</u>



## Why is **low-T** thermochronology useful?



• Low-temperature thermochronometers are unique because of their increased <u>sensitivity to topography</u>, <u>erosional and tectonic processes</u>



#### High temperature = no topography sensitivity

(a) High T<sub>c</sub> thermochronometers



- For thermochronometers with a high effective closure temperature, the closure temperature isotherm will not be influenced by surface topography
  - Note that age will increase with elevation as a result of the topography





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#### Low-temperature = sensitive to topography

(b) Low T<sub>C</sub> thermochronometry



- The effective closure temperature isotherm for lowtemperature thermochronometers <u>will generally be "bent" by</u> <u>the surface topography</u>, changing the age-elevation trend
  - The lower the value of T<sub>c</sub>, the more its geometry will resemble the surface topography





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- Because T<sub>c</sub> is sensitive to topography for low-temperature thermochronometers, it is possible to record changes in topography in the past (!)
  - Here, topographic relief decreases and the age-elevation trend gets inverted (older at low elevation)



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#### Common thermochronometers





- (U-Th)/He thermochronology is based on the production and accumulation of <sup>4</sup>He from parent isotopes <sup>238</sup>U, <sup>235</sup>U, <sup>232</sup>Th and <sup>147</sup>Sm
- <sup>4</sup>He ( $\alpha$  particles) produced during decay chains
  - $^{238}\text{U}$  8  $\alpha$  decays
  - $^{235}U 7 \alpha$  decays
  - $^{232}$ Th 6  $\alpha$  decays
  - $^{147}$ Sm I  $\alpha$  decay

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Ignoring the contribution of <sup>147</sup>Sm, we can say that the production of <sup>4</sup>He is

<sup>4</sup>He = 8 ×<sup>238</sup> U (e<sup>$$\lambda_{238}t$$</sup> - 1)  
+ 7 ×  $\frac{^{238}U}{137.88}$  (e <sup>$\lambda_{235}t$</sup>  - 1)  
+ 6 ×<sup>232</sup> Th (e <sup>$\lambda_{232}t$</sup>  - 1)

where <sup>4</sup>He, <sup>238</sup>U and <sup>232</sup>Th are the present-day abundances of those isotopes, *t* is the He age and the  $\lambda$  values are the decay constants

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Ehlers and Farley, 2003

Nice, datable apatites

Not-so-nice apatites

Ages are calculated by measuring the <sup>4</sup>He concentration by heating and degassing the mineral sample, then separately measuring the U and Th concentrations, for example by using an inductively coupled plasma mass spectrometer (ICP-MS)





 Selected mineral grains for dating should be high-quality, euhedral minerals free of mineral inclusions with a prismatic crystal form

- Why does the crystal form matter? Alpha particles travel ~20 µm when created and may be ejected from or injected to the sample crystal
- We can correct for this!

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## Fission-track dating - FT method





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Tagami and O'Sullivan, 2005

- Fission-track dating is based on measuring the <u>accumulation of damage</u> <u>trails in a host crystal</u> as the result of spontaneous fission of <sup>238</sup>U
  - Fission splits the <sup>238</sup>U atom into two fragments that repel and damage the crystal lattice over the distance they travel
  - In apatite, fresh fission tracks are ~16 µm long and ~11 µm long in zircon
- Similar to diffusive loss of <sup>4</sup>He, these <u>damage trails will be repaired, or anneal</u>, at temperatures above T<sub>c</sub>



### Fission-track dating - FT method



Tagami and O'Sullivan, 2005

- To be visible under a microscope, tracks must be chemically etched and enlarged
- At this point, tracks can be manually (or automatically) counted to determine the track density

$$t = \frac{1}{\lambda_{\rm D}} \ln \left( \frac{\lambda_{\rm D}}{\lambda_{\rm f}} \frac{N_{\rm s}}{^{238}{\rm U}} + 1 \right)$$

where  $\lambda_D$  is the <sup>238</sup>U decay constant,  $\lambda_f$ is the fission decay constant,  $N_s$  is the number of spontaneous fission tracks in the sample and <sup>238</sup>U is the number of <sup>238</sup>U atoms



- Argon dating is based on the decay of <sup>40</sup>K to radiogenic <sup>40</sup>Ar
  - Potassium is one of the most abundant elements in the crust, making argon dating one of the more common thermochronology methods
- <sup>40</sup>Ar/<sup>39</sup>Ar dating is used on white micas, biotite, K-feldspar and amphiboles



## Argon dating - <sup>40</sup>Ar/<sup>39</sup>Ar method

- <sup>40</sup>Ar/<sup>39</sup>Ar ages are found by <u>irradiating a sample (and standard)</u> with fast neutrons, producing <sup>39</sup>Ar from <sup>39</sup>K in the sample
- The <sup>40</sup>Ar/<sup>39</sup>Ar ratio is then measured as samples are either degassed entirely or step heated (next slide)
- The <sup>40</sup>Ar/<sup>39</sup>Ar age can be calculated as

$$t = \frac{1}{\lambda} \ln \left( 1 + J \frac{^{40}\text{Ar}}{^{39}\text{Ar}} \right)$$

where  $\lambda$  is the decay constant of <sup>40</sup>K, <sup>40</sup>Ar/<sup>39</sup>Ar is the measured sample <sup>40</sup>Ar/<sup>39</sup>Ar ratio and J is the irradiation factor  $J = \frac{e^{\lambda t} - 1}{\frac{40}{40} Ar/^{39} Ar}$ 

where *t* is a known age for a standard and <sup>40</sup>Ar/<sup>39</sup>Ar is its measured <sup>40</sup>Ar/<sup>39</sup>Ar ratio

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## Argon dating - Step heating



Harrison and Zeitler, 2005

- Step heating of <sup>40</sup>Ar/<sup>39</sup>Ar samples involves stepwise heating of samples to <u>gradually</u> <u>release Ar as the sample temperature</u> <u>increases</u>
- With this, it is possible to see the <sup>40</sup>Ar distribution in the sample, which is a function of the sample cooling history



## Argon dating - Step heating



- As we have seen on the previous slide,
  - (a) flat age spectra indicate rapid cooling of a rock sample (at time t<sub>1</sub>, here)
  - (b) spectra with lower concentrations initially either indicate partial reheating of the sample at time  $t_2$  or slow cooling from  $t_1$  to  $t_2$
  - (c) an unexpected behavior with higher Ar concentrations initially (i.e., near the rim of the grain)!
  - This "excess" Ar may have been taken up from surrounding minerals

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#### Common thermochronometers





• Why is low-temperature thermochronology a particularly interesting tool for those interested in geomorphology or active tectonics?

 How is are (U-Th)/He or <sup>40</sup>Ar/<sup>39</sup>Ar methods different from fission-track dating?



• Why is low-temperature thermochronology a particularly interesting tool for those interested in geomorphology or active tectonics?

 How is are (U-Th)/He or <sup>40</sup>Ar/<sup>39</sup>Ar methods different from fission-track dating?



Final project primer

- The final two exercises will be based on **thermochronology** 
  - The exercises will be <u>divided into two parts</u>, with the second exercise building on what you will have done the previous week
  - As usual, you will write/modify a Jupyter notebook code to produce some plots and provide short answers to some related questions
  - The questions you will answer for the write-ups for these two exercises will be relatively simple, only to let me know that you were able to do the requested tasks, because...



## Lab and final project primer

- ...you will expand on the work you do in the final two labs in a <u>formal written report</u>
- The report will be <u>no longer than 6-8 typed pages</u> (single spaced) including figures and references
- The idea is to describe some background on the data you will work with, the concept for its interpretation and your results/ conclusions
- The structure for the report is described in detail on the course webpage



#### References

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